



WELCOME!!

Presented by Steven Martinez



August 2017 CHAMP Program

Lecture Day 1 Cleanroom Fundamentals

Lecture Day 2 Characterization and Back End Processes

Lecture Day 3 Photolithography

Lecture Day 4 Deposition

Lecture Day 5 Etching

Lecture Day 1 Cleanroom Fundamentals

Intro to INRF & BiON at UCI

Example Applications

Intro to Cleanroom Design

Intro to Cleanroom Safety

Intro to Wafers and Starting Substrates

Intro to Deposition

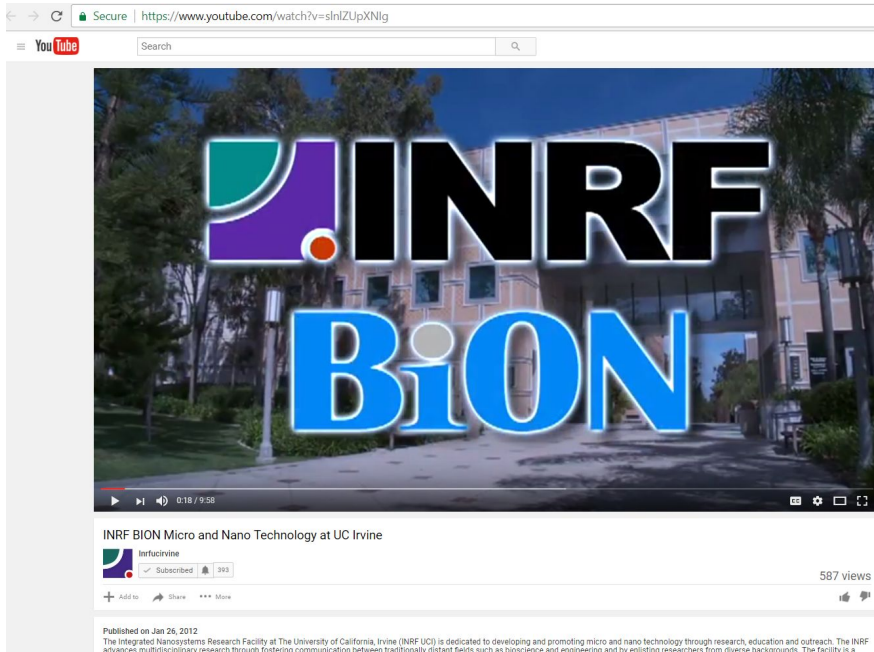
Intro to Photolithography

Intro to Etching (Wet and Dry)

Intro to Characterization

Intro to Back End Processes

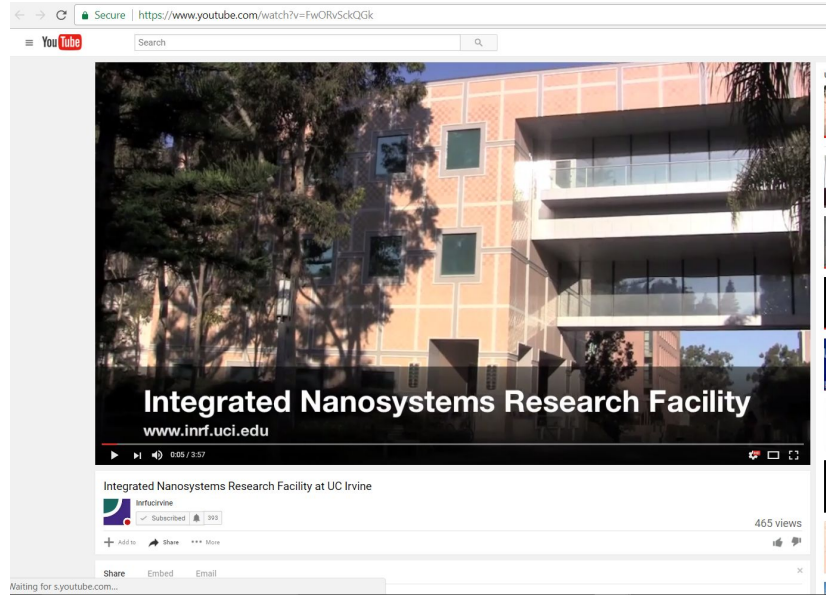
Intro to INRF & BiON at UC Irvine



The image is a screenshot of a YouTube video player. The browser's address bar shows the URL <https://www.youtube.com/watch?v=slnIZUpXNlg>. The video player itself shows a large, stylized logo for 'INRF BiON' overlaid on a photograph of a modern building at UC Irvine. The logo consists of a purple and teal graphic element followed by 'INRF' in large black letters and 'BiON' in large blue letters. Below the video player, the title 'INRF BION Micro and Nano Technology at UC Irvine' is visible, along with the channel name 'InfraUrvine', a 'Subscribed' button, and a view count of '587 views'. At the bottom of the player, there is a small text block: 'Published on Jan 26, 2012. The Integrated Nanosystems Research Facility at The University of California, Irvine (INRF UCI) is dedicated to developing and promoting micro and nano technology through research, education and outreach. The INRF advances multidisciplinary research through fostering communication between traditionally distant fields such as bioscience and engineering and by enlisting researchers from diverse backgrounds. The facility is a

<https://www.youtube.com/watch?v=slnIZUpXNlg>

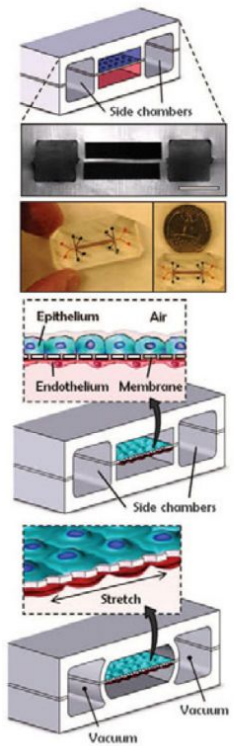
Intro to INRF & BiON at UC Irvine



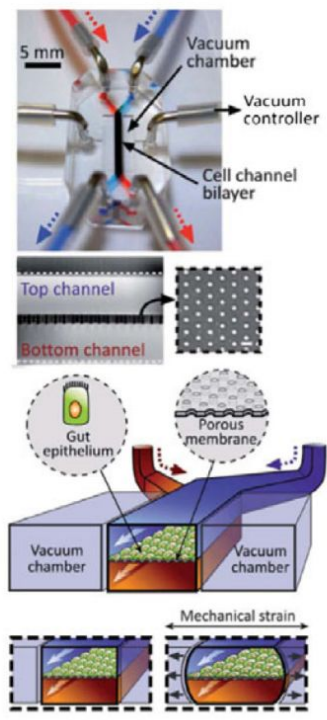
<https://youtu.be/FwORvSckQGk>

Example Applications

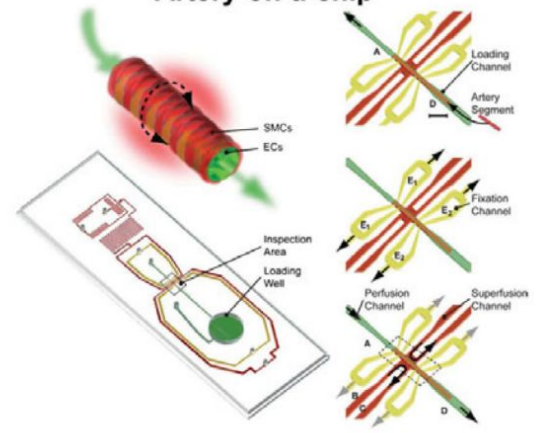
Lung-on-a-chip



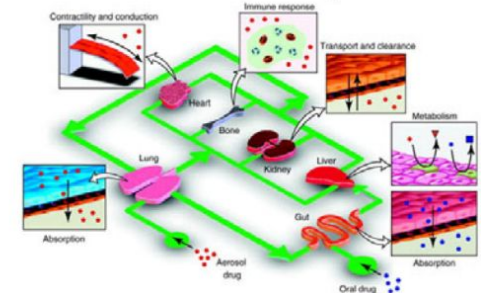
Gut-on-a-chip



Artery-on-a-chip



Human-on-a-chip





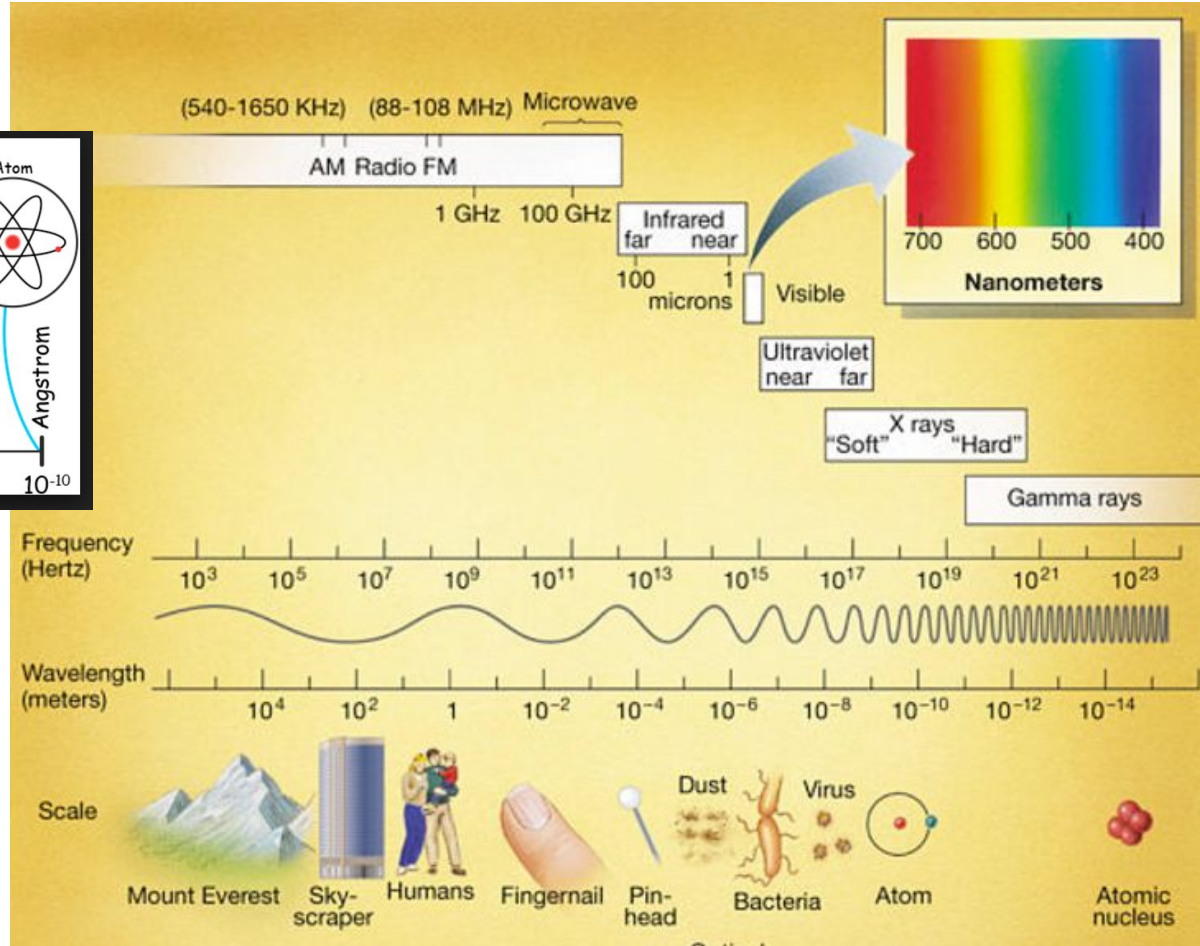
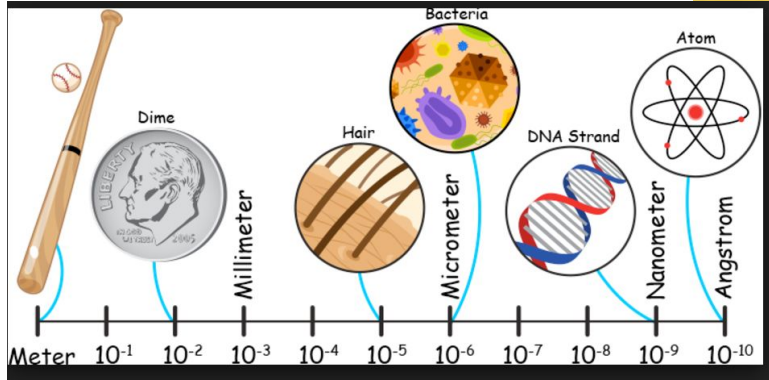
CLEANROOM FUNDAMENTALS



Intro to Cleanroom Design



Scale Reminder



Classification of Cleanroom

ISO 14644-1 and ISO 14698 [\[edit \]](#)

ISO 14644-1 and ISO 14698 are non-governmental standards developed by the [International Organization for Standardization \(ISO\)](#).^{[1][1]} The former applies to clean rooms in general (see table below); the latter to cleanrooms where biocontamination may be an issue.

Class	maximum particles/m ³ ^a						FED STD 209E equivalent
	≥0.1 μm	≥0.2 μm	≥0.3 μm	≥0.5 μm	≥1 μm	≥5 μm	
ISO 1	10 ^b	d	d	d	d	e	
ISO 2	100	24 ^b	10 ^b	d	d	e	
ISO 3	1,000	237	102	35 ^b	d	e	Class 1
ISO 4	10,000	2,370	1,020	352	83 ^b	e	Class 10
ISO 5	100,000	23,700	10,200	3,520	832	d,e,f	Class 100
ISO 6	1,000,000	237,000	102,000	35,200	8,320	293	Class 1,000
ISO 7	c	c	c	352,000	83,200	2,930	Class 10,000
ISO 8	c	c	c	3,520,000	832,000	29,300	Class 100,000
ISO 9	c	c	c	35,200,000	8,320,000	293,000	Room air

^a All concentrations in the table are cumulative, e.g. for ISO Class 5, the 10 200 particles shown at 0.3 μm include all particles equal to and greater than this size.

^b These concentrations will lead to large air sample volumes for classification. Sequential sampling procedure may be applied; see Annex D.

^c Concentration limits are not applicable in this region of the table due to very high particle concentration.

^d Sampling and statistical limitations for particles in low concentrations make classification inappropriate.

^e Sample collection limitations for both particles in low concentrations and sizes greater than 1 μm make classification at this particle size inappropriate, due to potential particle losses in the sampling system.

^f In order to specify this particle size in association with ISO Class 5, the macroparticle descriptor M may be adapted and used in conjunction with at least one other particle size. (See C.7.)

^g This class is only applicable for the in-operation state.

US FED STD 209E [\[edit \]](#)

US FED STD 209E was a [United States federal standard](#). It was officially cancelled by the [General Services Administration](#) on November 29, 2001,^{[12][13]} but is still widely used.^[14]

Class	maximum particles/ft ³					ISO equivalent
	≥0.1 μm	≥0.2 μm	≥0.3 μm	≥0.5 μm	≥5 μm	
1	35	7.5	3	1	0.007	ISO 3
10	350	75	30	10	0.07	ISO 4
100	3,500	750	300	100	0.7	ISO 5
1,000	35,000	7,500	3000	1,000	7	ISO 6
10,000	350,000	75,000	30,000	10,000	70	ISO 7
100,000	3.5 × 10 ⁶	750,000	300,000	100,000	700	ISO 8

Cleanrooms

American Physicist Willis Whitfield at Sandia Laboratories invented the modern cleanroom in 1960.

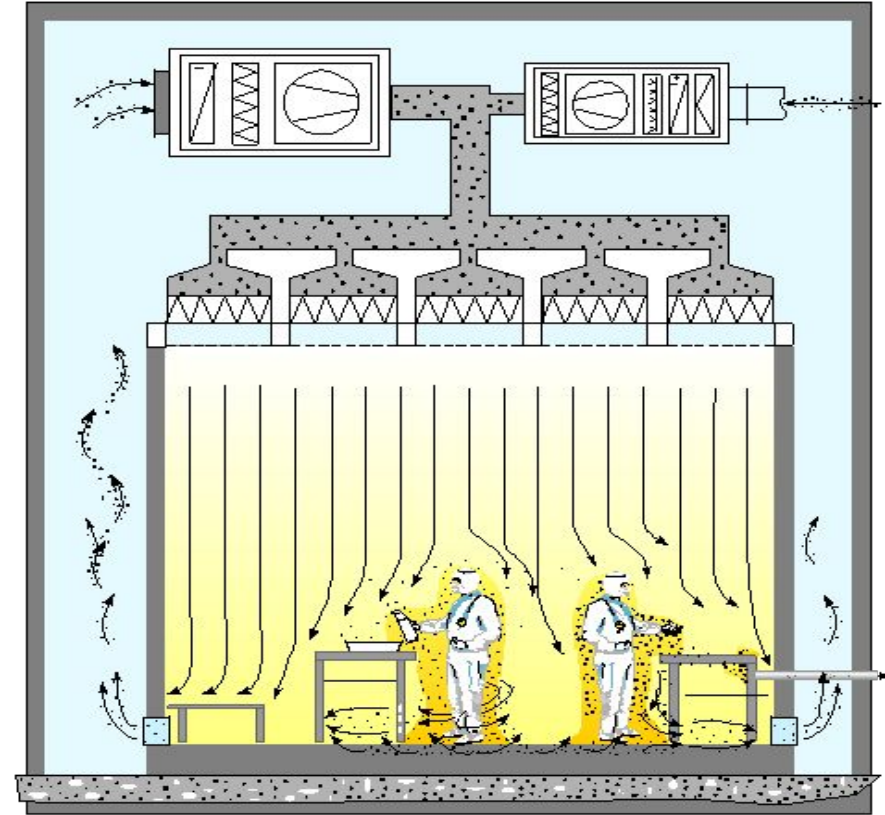
A cleanroom is typically used in semiconductor manufacturing, scientific research, biotechnology, hospitals, pharmaceutical industries, and other industries sensitive to particles contamination. (Ex. NASA Facilities, INTEL manufacturing, universities, biotech companies, etc.)

Cleanrooms have generated over 50 billion dollars in sales and entire manufacturing facilities contain a cleanroom environment covering thousands of square meters.

Cleanrooms use HVAC systems to pull outside air into the room and filter the intake air in the first stage by an Air pre-filter and the second stage air is filtered through a filter (called a HEPA Filter).

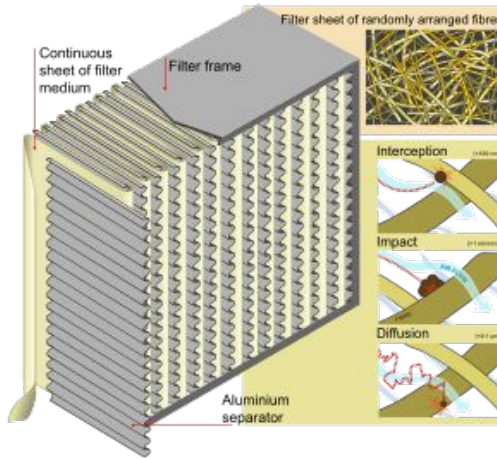
The pre-filter in the first stage removes particles 25 microns or greater. The second stage HEPA Filter removes particles from 0.3 micron to 25 microns.

In ambient outside air there are over 35,000,000 particles in the size range from 0.5 microns and greater in diameter.

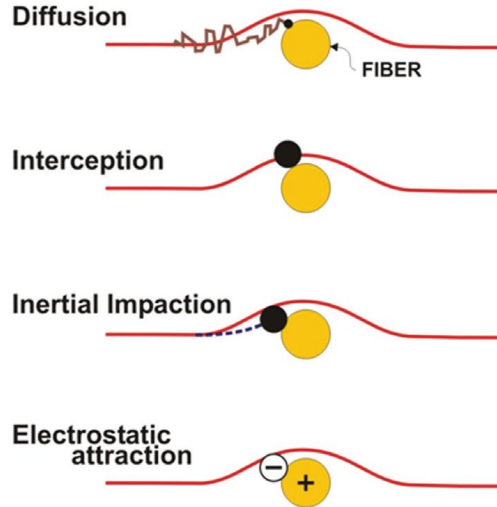


What is HEPA Filter

- High Efficiency Particle Arrestor (HEPA) can trap particles upto 0.3 microns or even smaller – Traps the tiniest of pollutants
- The quality, efficiency and life of a HEPA filter largely depends on the thickness, folded area and the material used in it

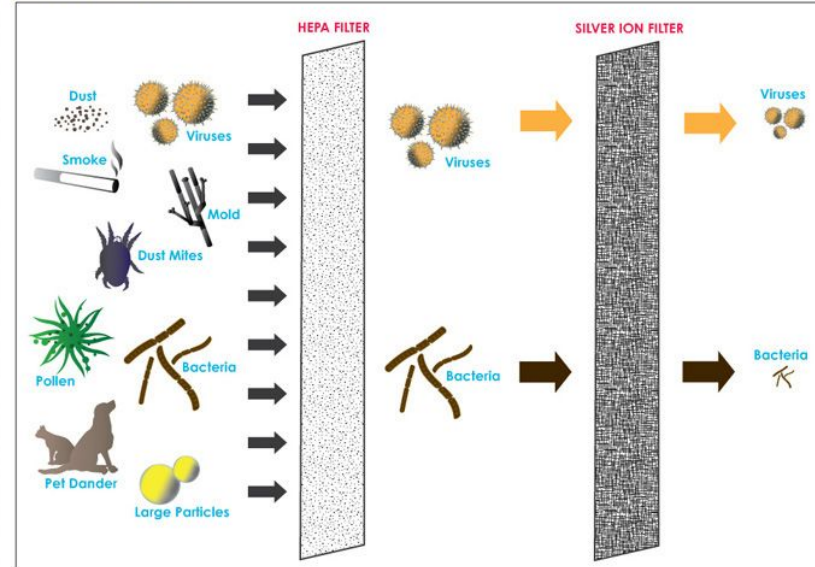


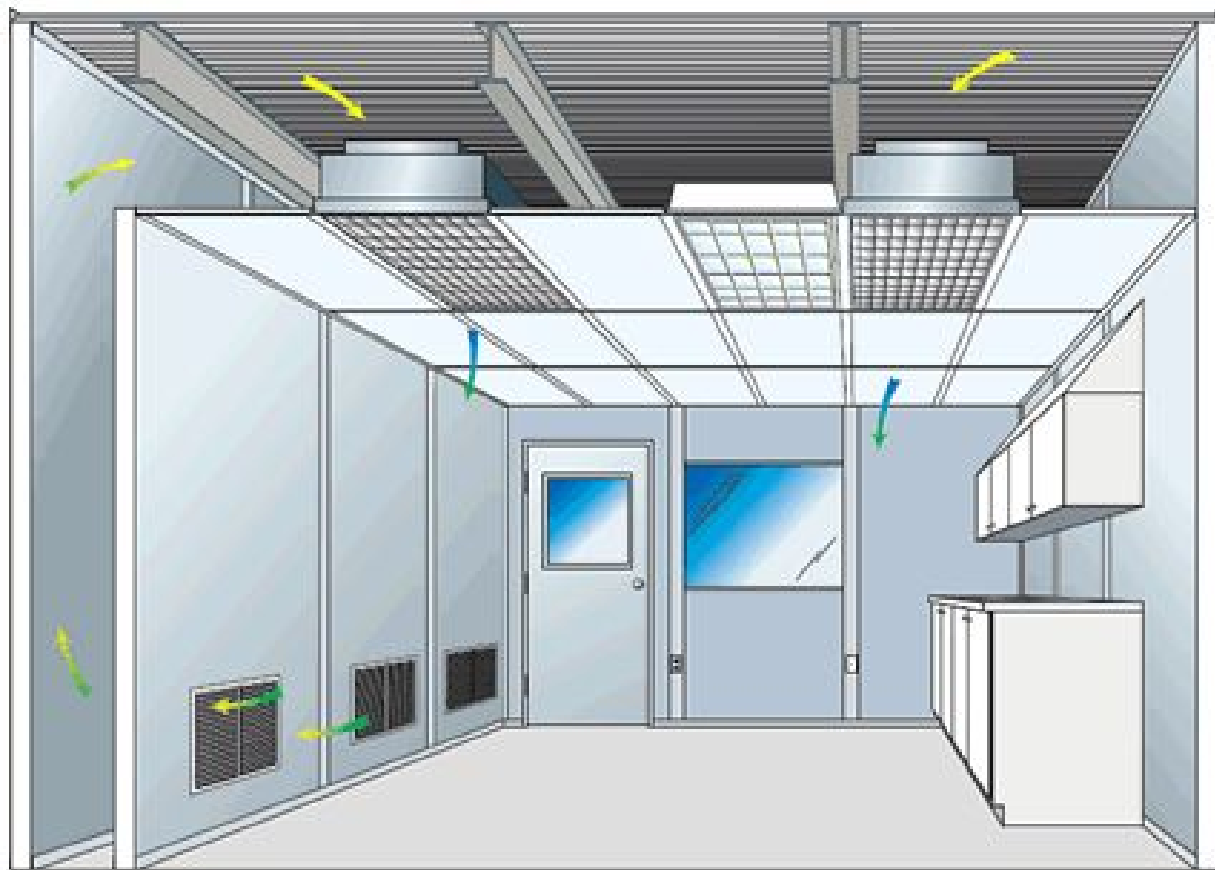
Filtration Mechanisms

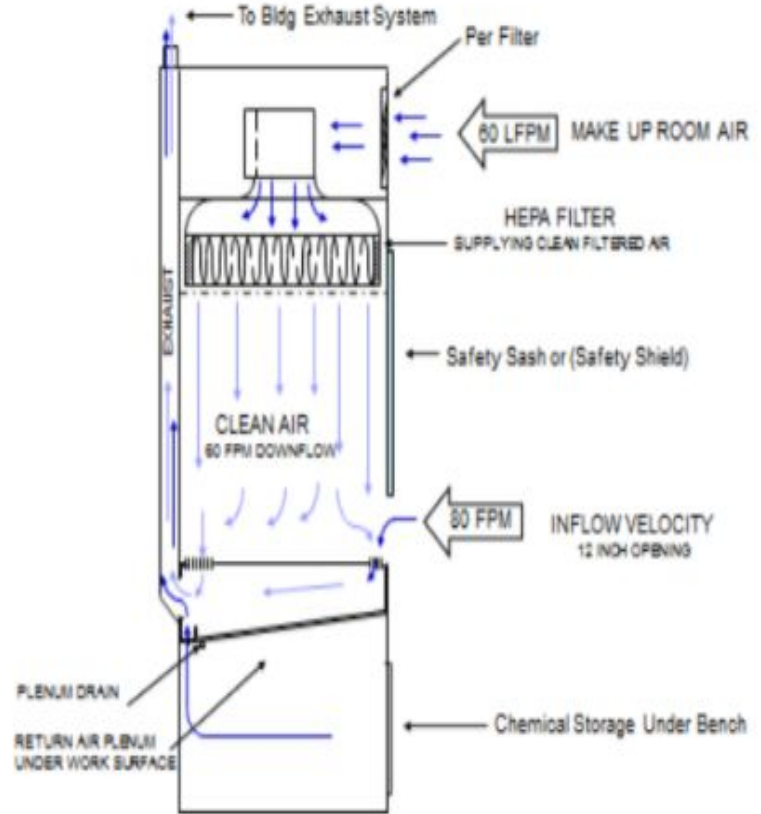
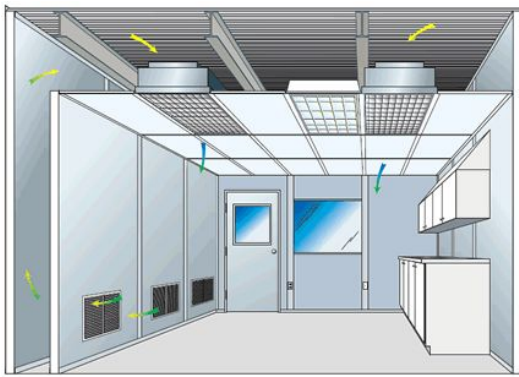


HEPA with Silver Ion Filtration

The HEPA filter eliminates over 99% of airborne allergens while the addition of the Silver Ion filter eliminates 98% of bacteria and half of airborne viruses.





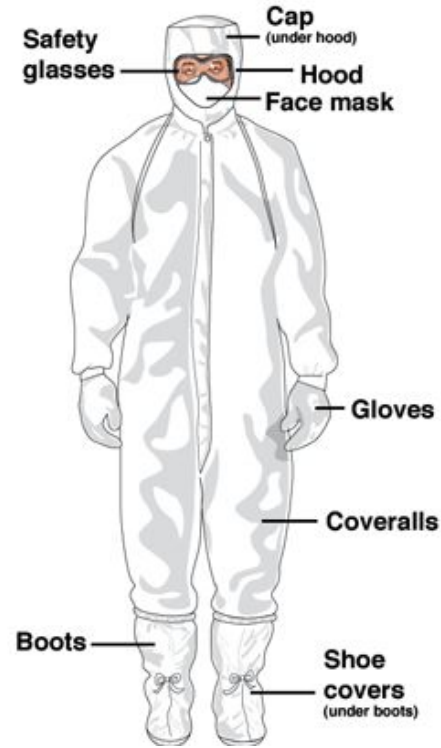


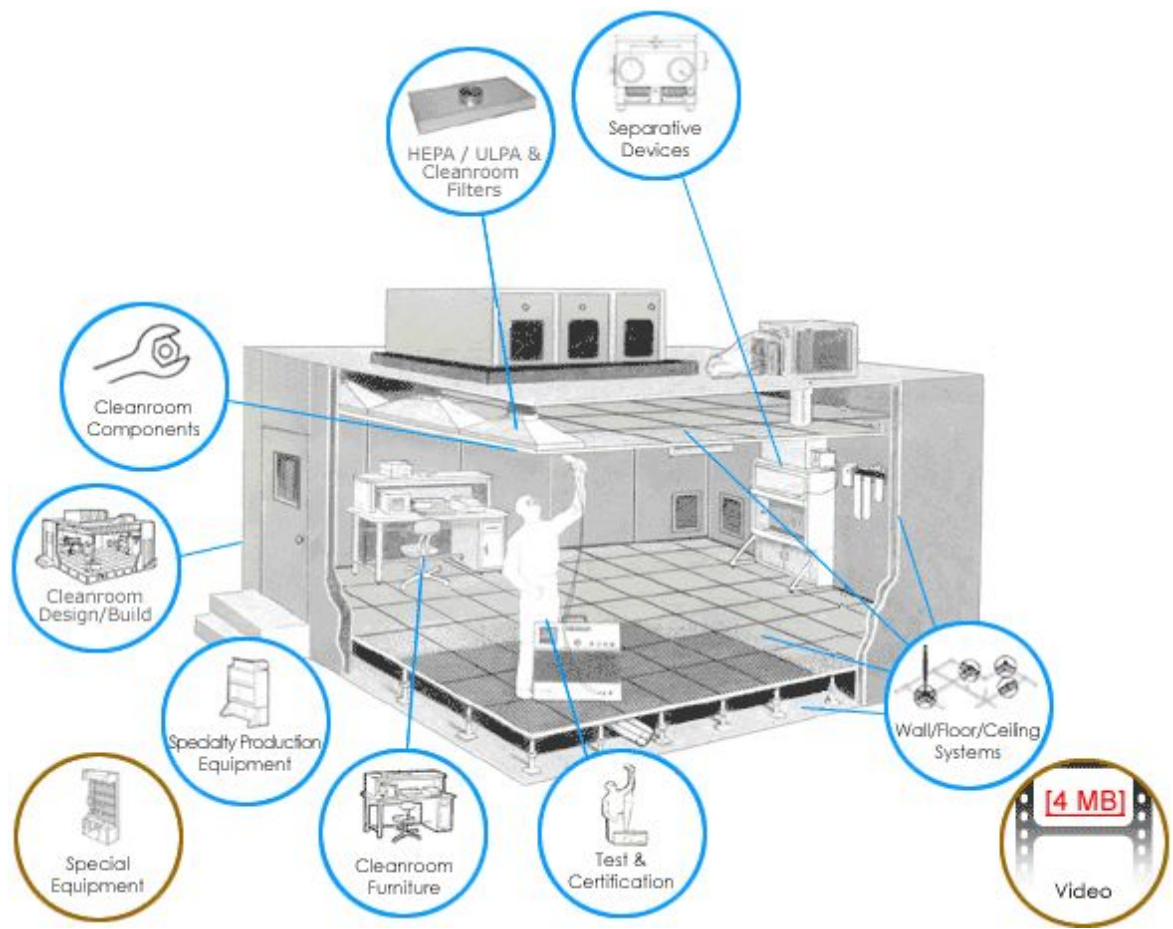
Very special protocol has to be followed once inside the cleanroom in order to maintain cleanliness classification.

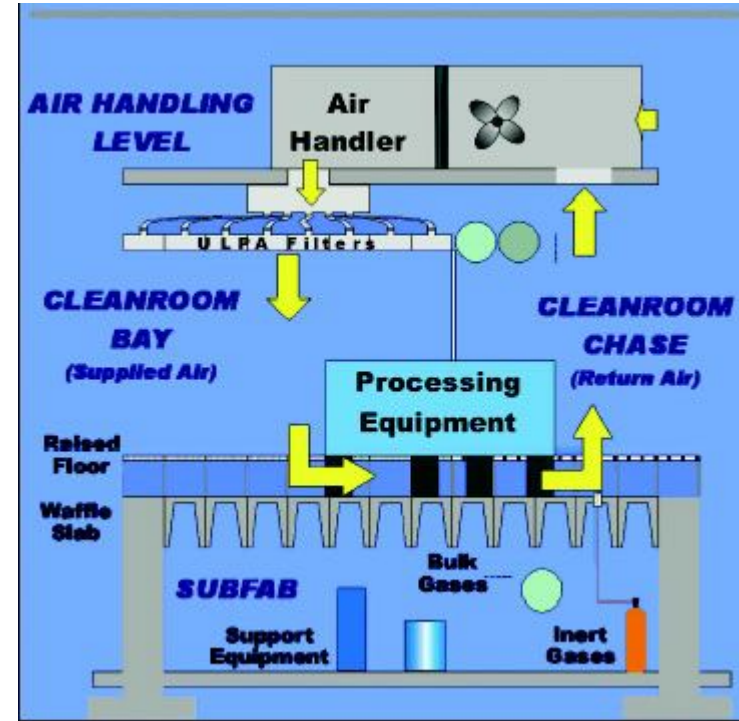
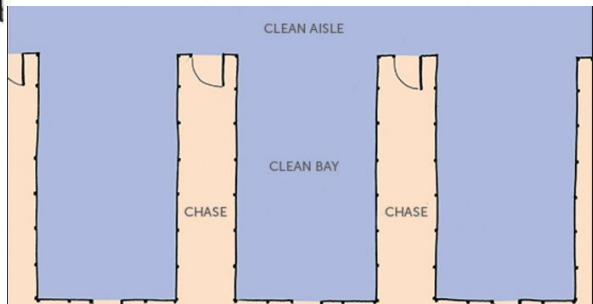
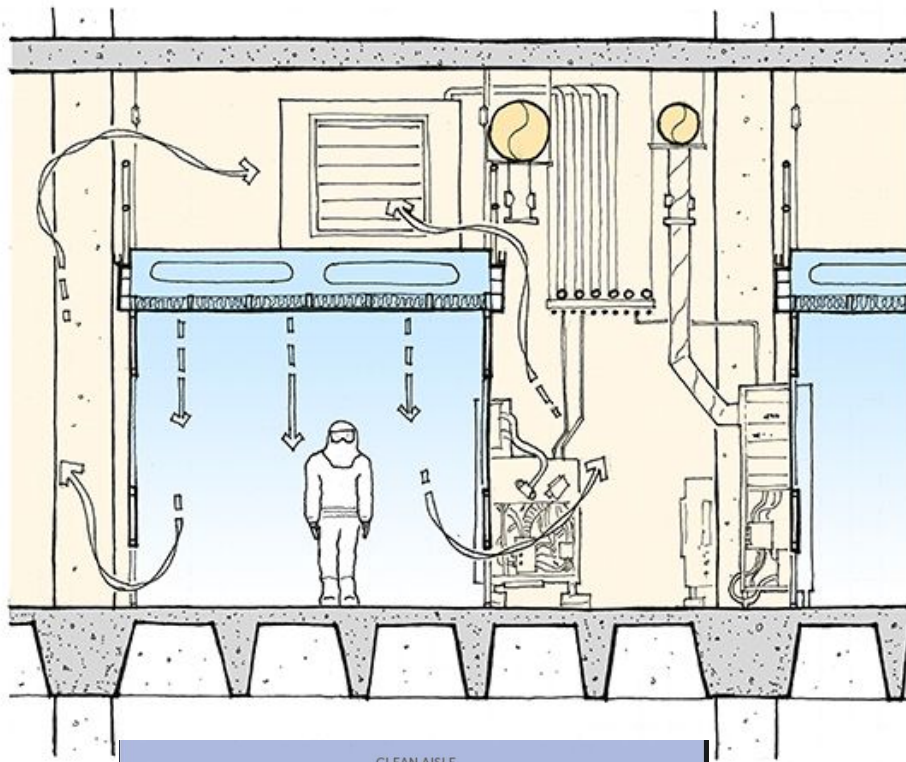
The user of the room must wear **protective** clothing such as coveralls, hoods, boots, masks, and gloves.

No paper and fabric materials are allowed in the room.

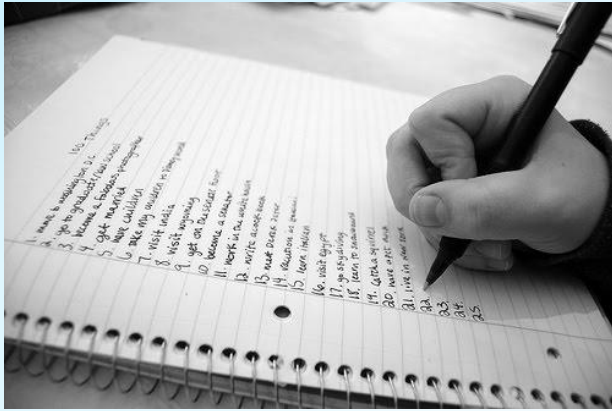
The water is purified for contamination and bacteria for maintaining the yield in processes performed in the cleanroom.







Safety in Cleanrooms



* There are a lot of important things to consider when planning a safe cleanroom environment.

People

- Irreplaceable
- Health (short term and long term)
- Customers (should feel safe)
- Trained
- Access to knowledge / resources



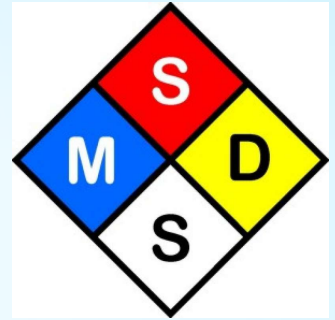
Facility

- Safe environment for people to conduct research/work/etc.
- Policies and Regulations of the University, the State, Federal
- Secure
- Equipment uptime
- Health of equipment; minimize cost of repairs to damage caused by lack of safety

- Appropriate PPE (Goggles, Glasses, Gloves, Face Shields, Aprons, etc.)

*Bunny suits protect the cleanroom from the human, they DO NOT protect the human inside them

- Chemical Handling (Transport, Incoming, Storage, Use, Waste/Disposal)



- Prevention (Procedures, Posted Notices/Signs, Training)

- Response (Procedures, Posted Notices/Signs, Training)

- Knowledge (Regular Training, Literature Access(MSDS, Contact Info, etc.), Open Dialogue...)

Miscellaneous Hazards: Sharps and Heavy Objects, Pinch Points, Hot/Cold Surfaces, Solid/Liquid/Gas Hazards, Flammability, Explosion, Implosion, Asphyxiates, Corrosion, UV Exposure, RF Exposure
Electrical Shock, Biohazards, Electromagnets, Cryogenics, Earthquake

Waste/Disposal: Toxic Waste Disposal, Decomposition/Oxidation, Wet Scrubbing Systems, Pollution Control

Resources: Accessibility (hard copies of MSDS in the lab), Safety Training, Hospital Info, Emergency Contacts

Knowledge: Safety Manuals, Postings, Certifications (Quizzes, Courses), Refresher Training, Evacuation Plan

Chemistry: Use of Hydrofluoric (HF) Acid and other hazardous chemicals, Chemicals being brought into the lab, Groups/users sharing bottles/beakers/waste containers, Proper Drainage, Compatibility with Labware, Proper Labelling (pre, during, and post use), Separation of Acids & Bases, Oxidizers

Other: PPE, Lockout/Tagout Procedures, Buddy Systems, Medical Emergencies, Relationship and Collaboration with EH&S department and other safety departments, Health Conditions (Pacemakers, Reproductive Health/Pregnancy Concerns, Allergies), Gas Cylinder Change Outs, Smells, Toxicity, Work benches, Exhaust, Air Flow, Lab Tidiness/Upkeep, Eyewashes & Showers, Fire Extinguishers, Smoke Detectors, Toxic Gas Detector Systems, Emergency Exits, Alarms (Sight and Sound), Proper Attire

Hydrofluoric Acid (HF)

Safety Awareness

Hydrofluoric acid (HF) is a clear and colorless corrosive liquid.

Though a weaker acid, HF is highly potent and physiologically active chemical.

- As little as 7 milliliters (mL) of anhydrous HF in contact with the skin and untreated can bind all the free calcium in a normal sized adult.*
- With burns involving greater than 25 square inches there is a significant risk for serious and potentially fatal systemic toxicity.*
- **There is no concentration of HF would can be relied upon as safe!**
- **There is no material that is completely resistant to HF degradation.**

* from: Baird D. and Cooper C., “Hydrofluoric Acid Safety”,

Fermilab GHS/RCR Hazards



Acute toxicity (H300,H310,H330): Fatal if swallowed, in contact with skin, or if inhaled.



Skin corrosion (H314): Causes severe skin burns and eye damage.
Serious eye damage (H318): Causes serious eye damage.

NFPA Fire Diamond



- ◆ Health (4): Can be lethal.
- ◆ Flammability (0): Will not burn under typical fire conditions.
- ◆ Instability (1): Normally stable, but can become unstable at elevated temperatures and pressures.
- ◇ Special: No NFPA 704 special hazards.

How is HF different from other acids?

Strong, inorganic acids (such as hydrochloric or sulfuric acid) on contact with live tissue produce immediate necrosis. Similarly, exposure to concentrated solutions of HF will produce significant tissue damage at the exposure site.

Additionally:

- HF is readily absorbed through the skin.
- Fluoride ions move rapidly through the skin causing deep tissue damage.
- HF binds to the available calcium and magnesium in the body. Depletion of the calcium reservoirs can induce hypocalcemia.
- Potassium ions (K^+) enter the extracellular space to compensate for the loss of the calcium.
- Nerve endings are irritated by the altered K^+ levels and produce severe pain.

Safe Work Practices

- Never use HF alone (buddy system).
- Training for both the hazards of HF and the manner in which it will be used:
 - Approved Standard Operating Procedures (SOPs)
 - Enhance with Job Safety Analyses (JSA) or Risk Assessments
- Always wear appropriate Personal Protective Equipment (PPE).
- Always use HF in a laboratory fume hood.
- Confirm that there is antidote (calcium gluconate) before handling HF.
- Use signs to designate areas and provide warning where HF will be in use.
- Use only with appropriate containers for handling HF reagents and wastes and accurately label the containers

Recommended PPE

- Long Neoprene gloves (avoid exposed wrists)
- Saranex coated coveralls or other chemically resistant body protection
- Face shield and goggles (acid resistant recommended)



Baird D. and Cooper C., "Hydrofluoric Acid Safety", Fermilab.

Container Resistance to HF

Not resistant:

Do not store HF in glass containers. Hydrofluoric acid reacts with many materials. Use one of the two recommended types of storage containers to store HF reagents and wastes.

Best resistance:

Teflon (PTFE)

- HF(48%) at 70°F for 6 months
- No notable change or deterioration of the container.

Moderate resistance:

High Density PolyEthylene (HDPE containers)

- HF(48%) at 70°F for 30 days
- Less than 15% swelling, less than 20% loss in tensile strength, and only minor evidence of chemical attack.

All primary HF containers must be labeled and in secondary containment.

Container Labels and Warning Signs


Any container that holds an HF solution should be clearly labeled.


Label should include:

- Chemical Name
- Associated Hazards
- Date created
- Recommended: Use a GHS formatted label

Hydrofluoric acid

DANGER

 Fatal if inhaled, if swallowed, or in contact with skin. Causes severe skin burns and eye damage. Effects may be delayed. Do not eat, drink or smoke when using this product. May be corrosive to metals.

 **PREVENTION**
Do not breathe the mist or vapors. Use only outdoors or in a well-ventilated area. Where exposure limits are exceeded, wear respiratory protection. Do not get in eyes, on skin, or on clothing. Wash skin and eyes thoroughly after handling. Wear protective gloves and clothing, and eye and face protection. Keep only in original container.

RESPONSE
Immediately call a doctor or other medical personnel.
If swallowed: Rinse mouth. If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. **If on skin (or hair):** Take off immediately all contaminated clothing. Wash skin with water or shower. Wash contaminated clothing before reuse. Contaminated skin must be treated with calcium gluconate solution. **If inhaled:** Remove person to fresh air and keep comfortable for breathing. Absorb spillage to prevent material damage.



Place labels on areas where HF solutions are used or being stored.

DANGER!

Hydrofluoric acid in use.

Acutely Toxic.

Corrosive.

What to do if an HF exposure occurs:

- Do not panic!
- Provide assistance
 - Help exposed individual to eyewash/safety shower
 - Take care to avoid spreading the contamination to unexposed parts or the assistant
 - Flush the exposed area with water for 5 minutes
 - Remove any contaminated clothing while under the shower
- Administer HF antidote (calcium gluconate)
- Call for emergency assistance and seek medical attention

Remember:

The best treatment for exposures is prevention.

- Explore the possibility of eliminating or substituting the HF for a less hazardous material
- Properly use Engineering Controls when handling HF
- Develop effective administrative controls for handling HF
- Use appropriate PPE when handling HF
- Continue to develop and refine your procedures and training practices when procedures change or when improvements are found

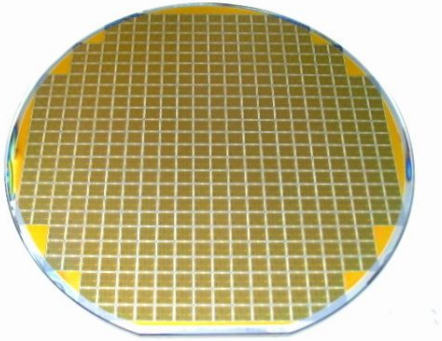
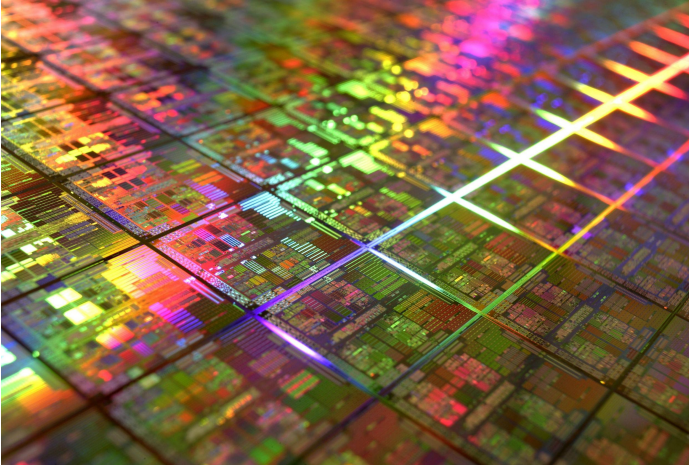
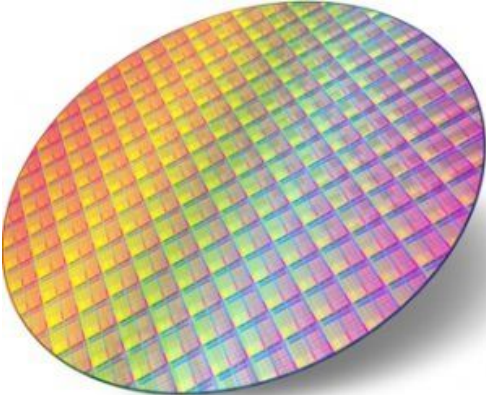
Back to the research.....

Semiconductors and Biotechnology research cleanrooms facilities utilize Photolithography, Dry etching, wet etching, Metal Deposition, Dielectric Deposition and back-end packaging rooms to perform fabrication-processing steps.

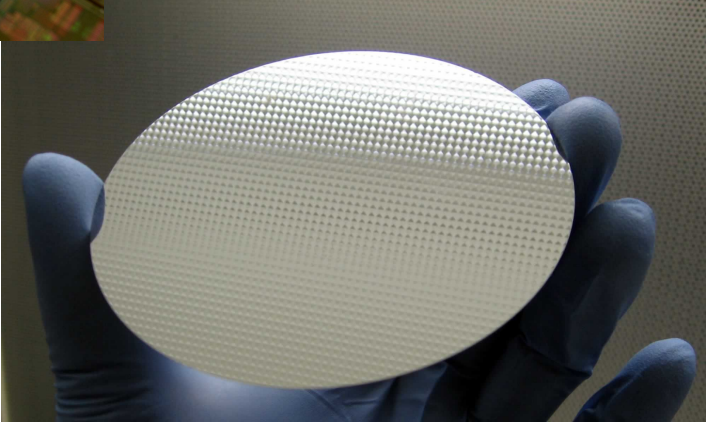
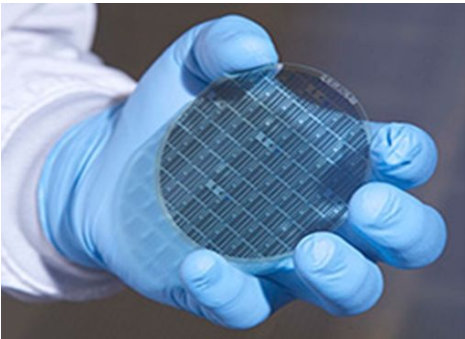
The UCI INRF and BION have 16,000 square feet of cleanroom space dedicated for semiconductor and biomedical device processing.

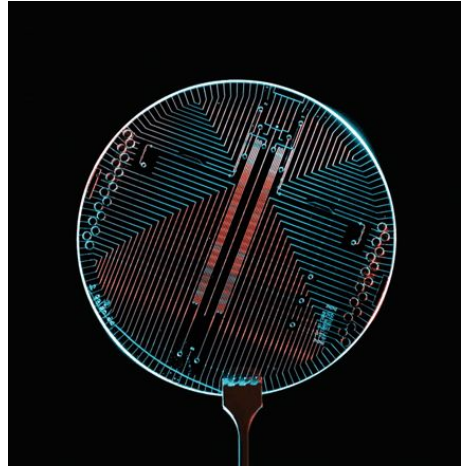
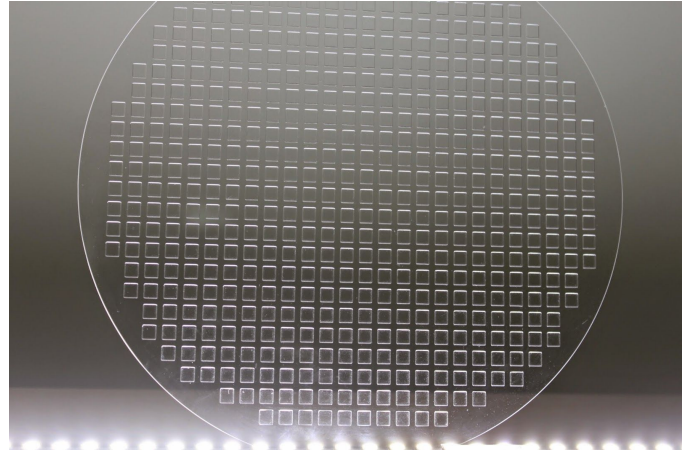
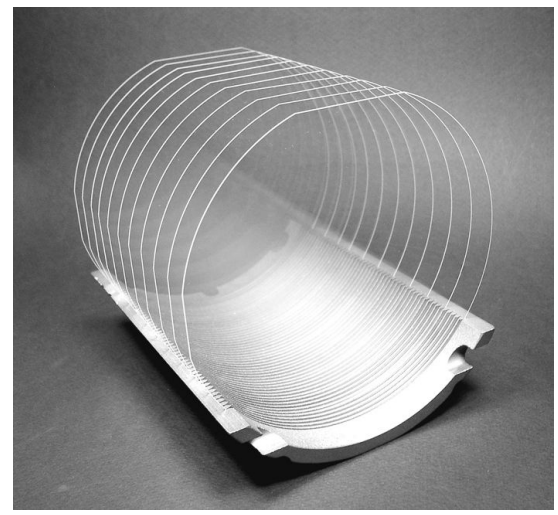
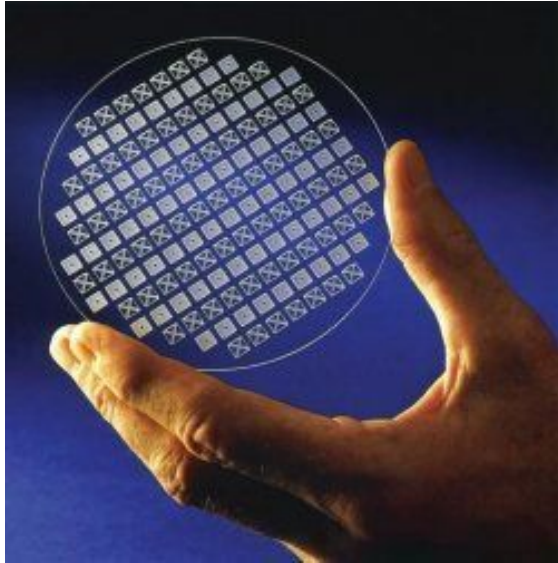
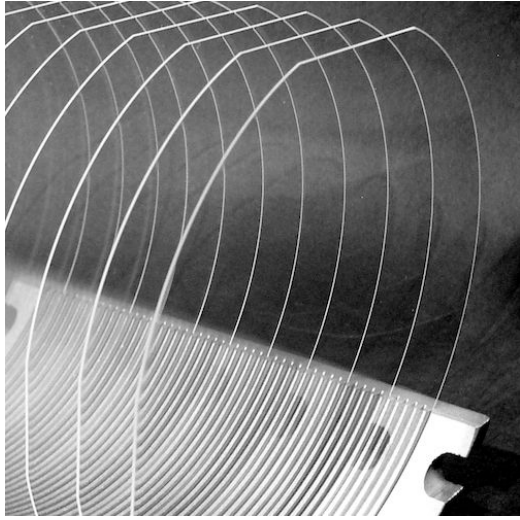
<http://www.inrf.uci.edu/orientation>

Intro to Wafers and Starting Substrates

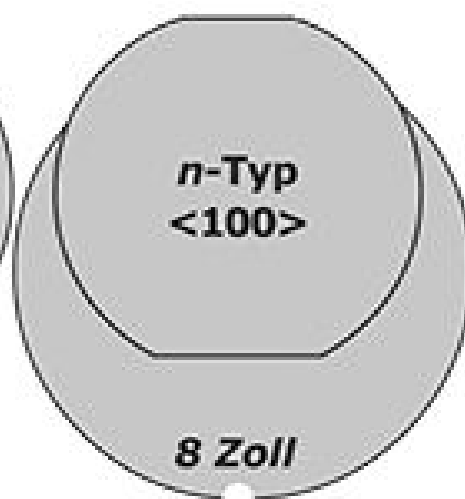
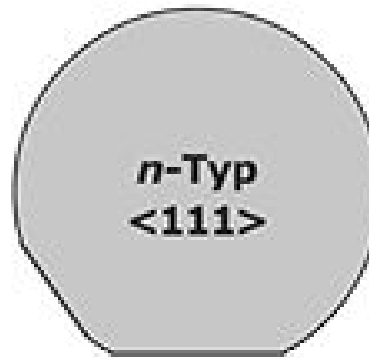


© ChipsEtc.com

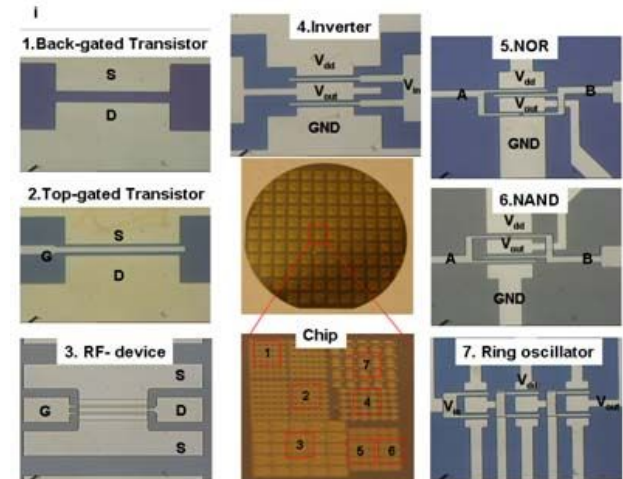
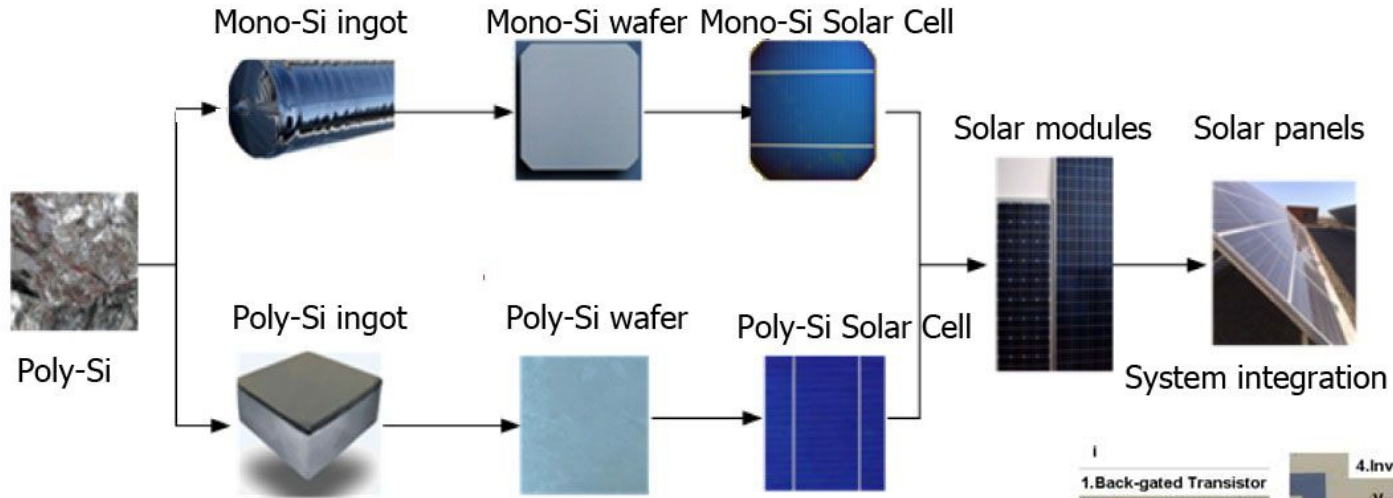


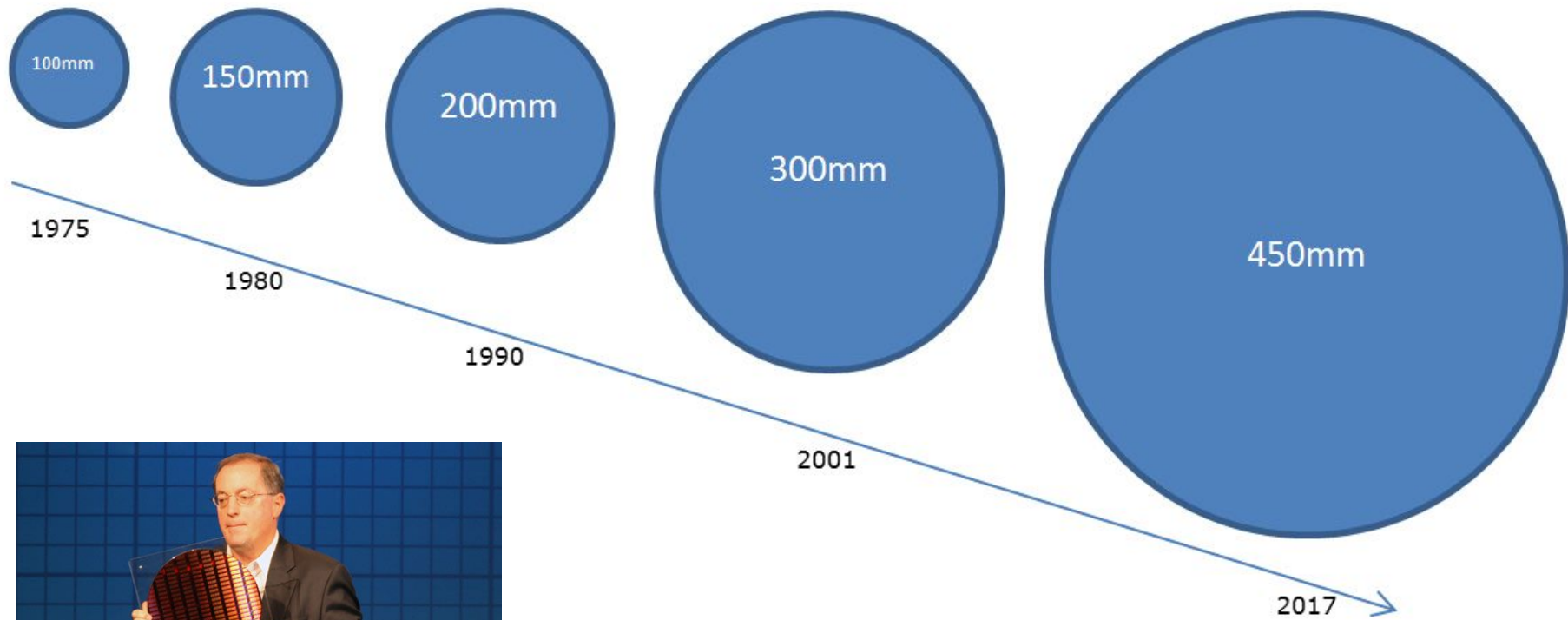






Solar grade wafers for PV application





Czochralski Growth



Preparation of Silicon Wafer

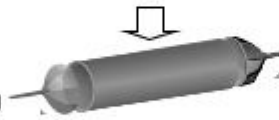
1. Crystal Growth



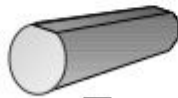
2. Single Crystal Ingot



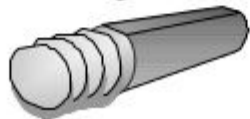
3. Crystal Trimming and Diameter Grind



4. Flat Grinding



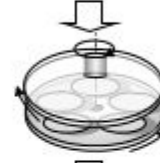
5. Wafer Slicing



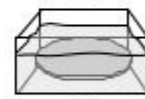
6. Edge Rounding



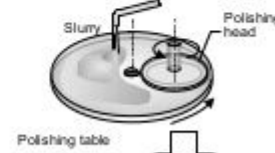
7. Lapping



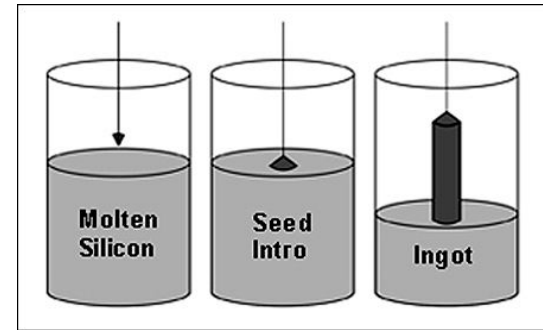
8. Wafer Etching

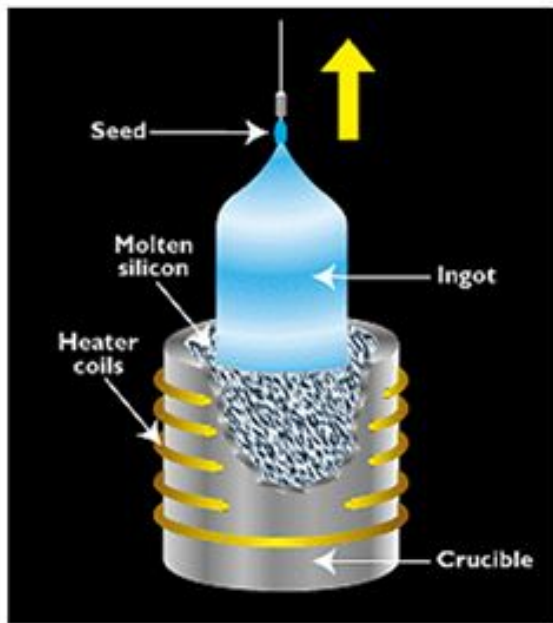
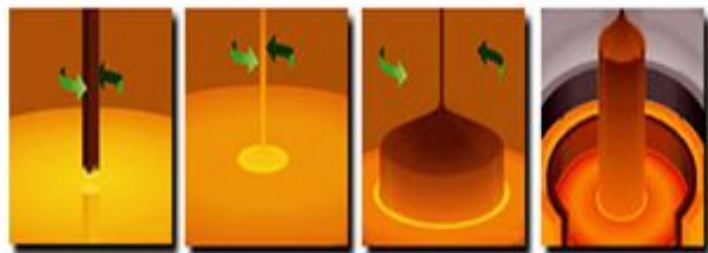


9. Polishing



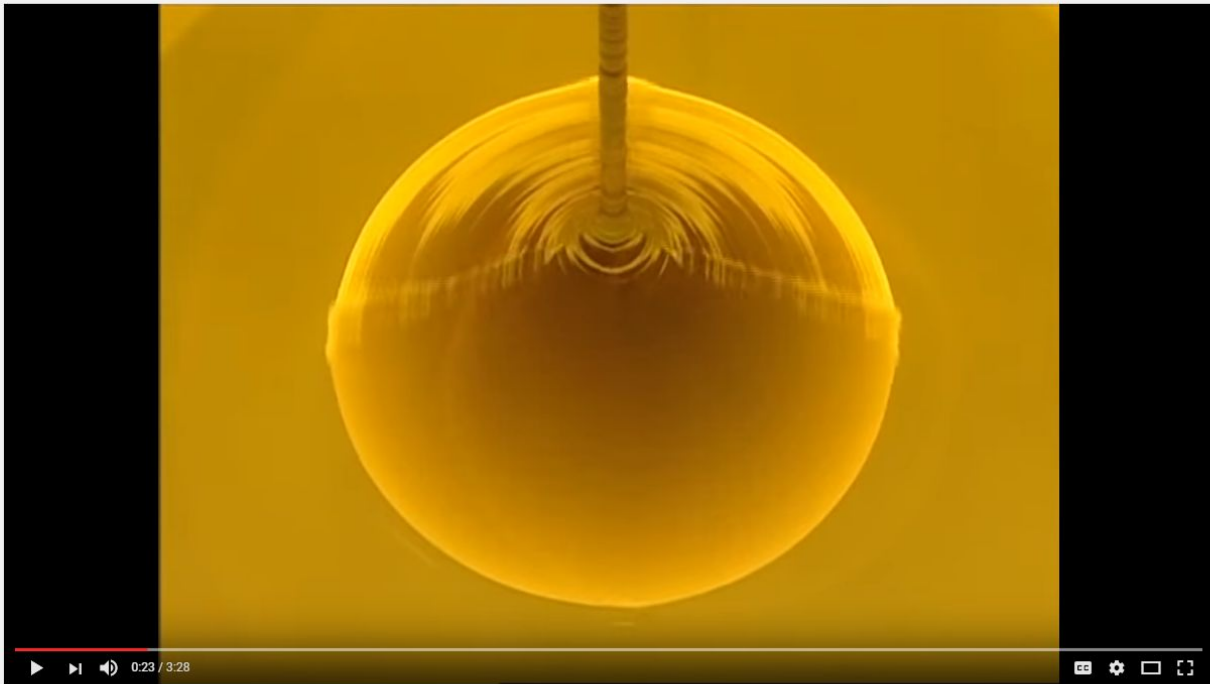
10. Wafer Inspection





→ ↻ Secure | <https://www.youtube.com/watch?v=jh2z-g7GJxE>

☰ YouTube | seed crystal wafer si 🔍



Up next

0:23 / 3:28

From sand to silicon

The image shows a YouTube video player interface. At the top, the browser's address bar displays the URL 'https://www.youtube.com/watch?v=jh2z-g7GJxE'. Below the address bar is the YouTube logo and a search bar containing the text 'seed crystal wafer si'. The main video player area shows a large, glowing yellow sphere, which is a seed crystal wafer, being rotated. The video player includes standard controls like play, volume, and a progress bar showing 0:23 / 3:28. Below the video player, the title 'From sand to silicon' is displayed. On the right side of the player, there is a vertical list of 'Up next' video thumbnails, including one with the Intel logo.

3min: <https://www.youtube.com/watch?v=jh2z-g7GJxE>

Intro to Deposition

Intro to Photolithography

Intro to Etching (Wet and Dry)

Intro to Characterization

Intro to Back End Processes

Intro to Semiconductor Processing

Quick Version

2min: <https://www.youtube.com/watch?v=JDROPMoNZpk>



Intro to Deposition → **Adding Material**

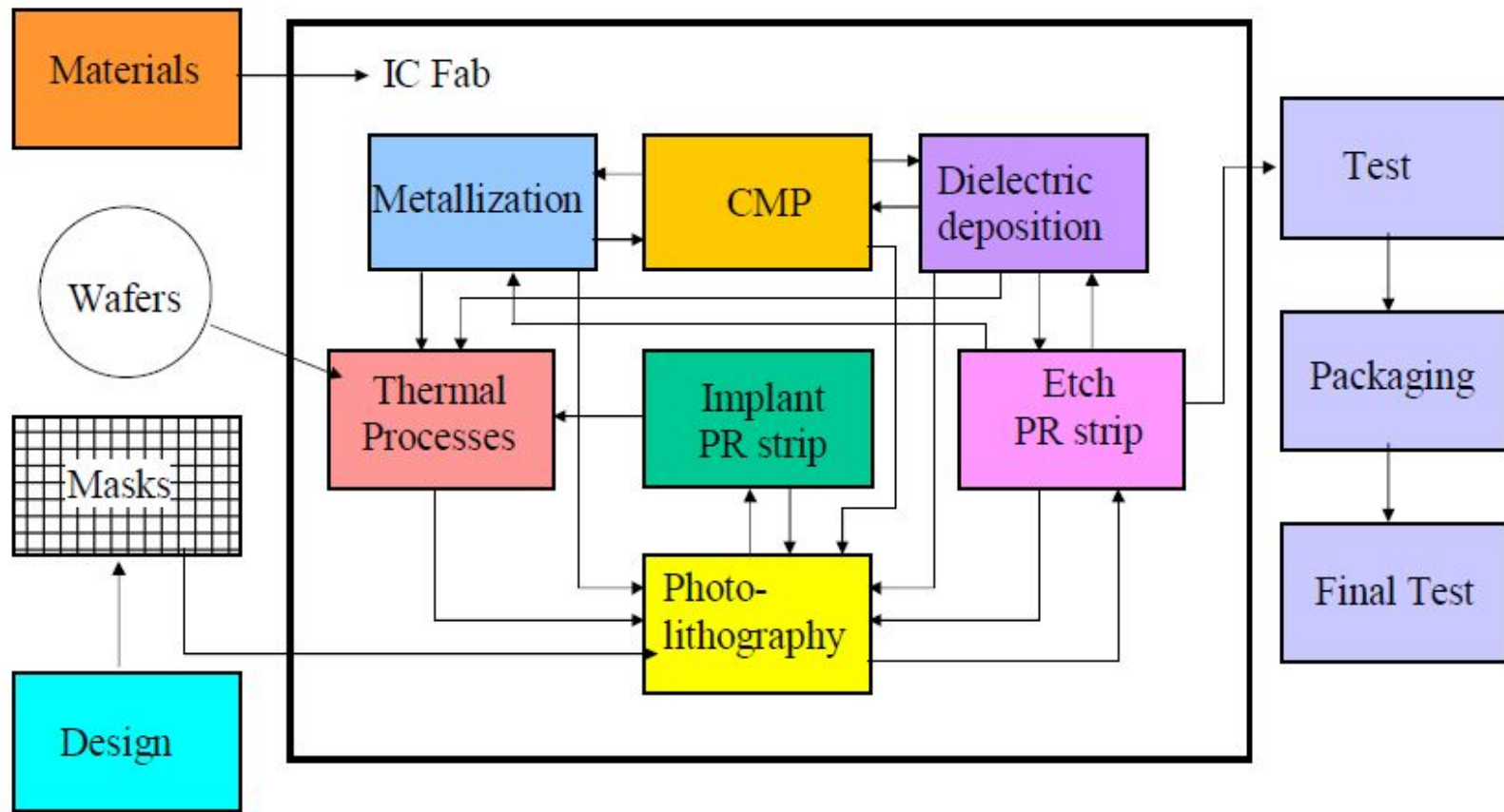
Intro to Photolithography → **Transferring Pattern**

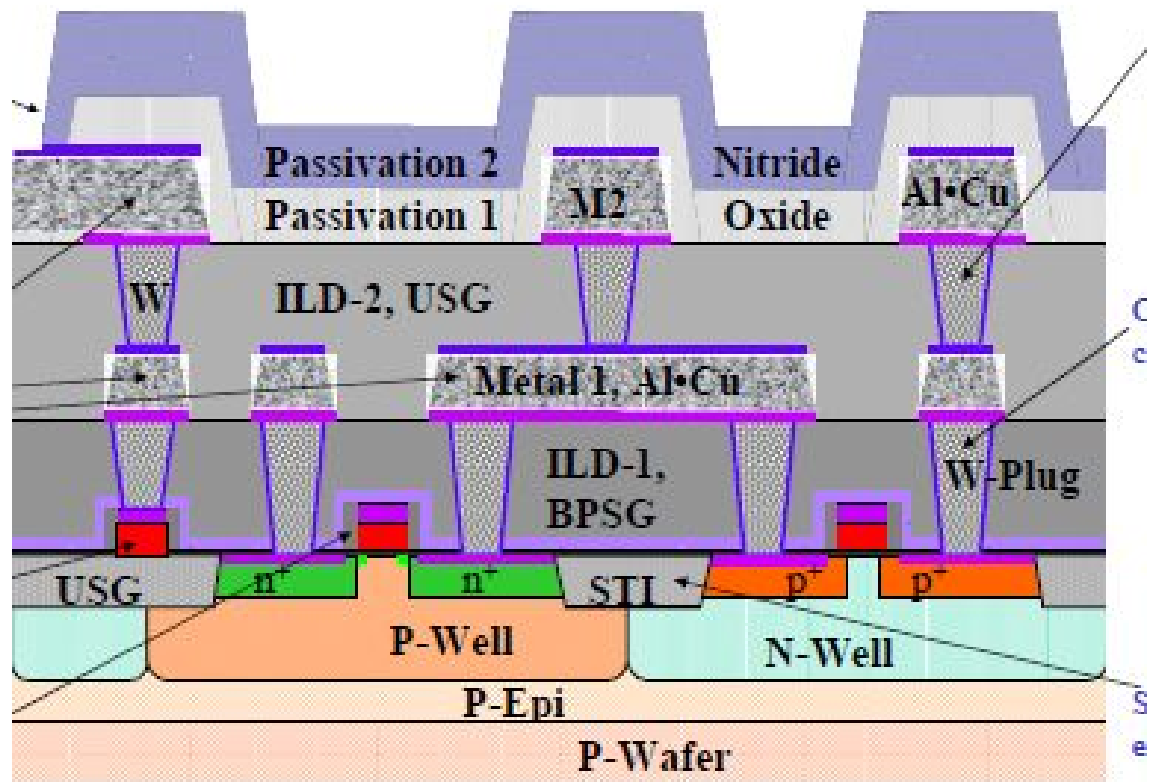
Intro to Etching (Wet and Dry) → **Removing Material**

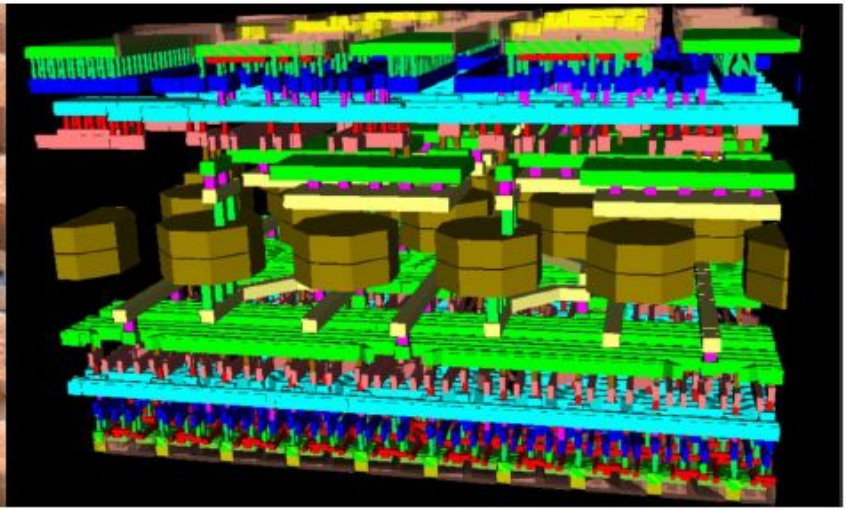
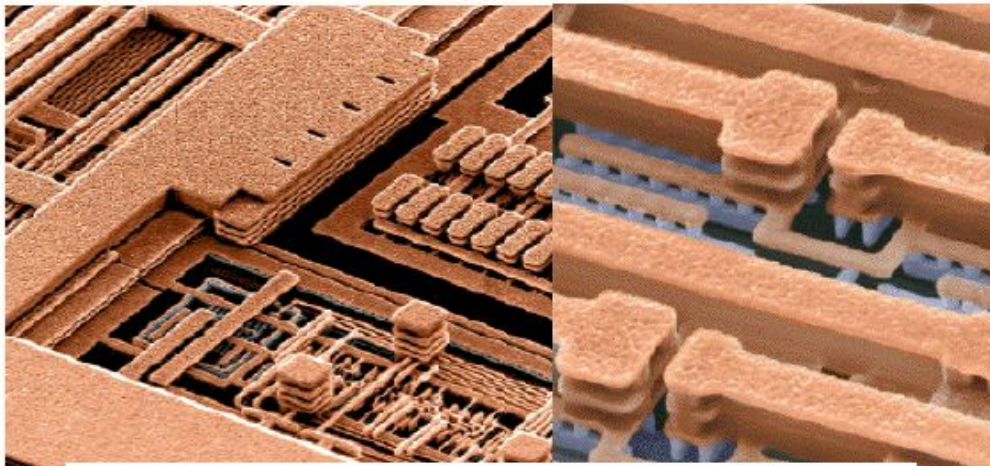
Intro to Characterization → **Observing and/or Measuring**

Intro to Back End Processes → **Final Steps, Transfer from Wafer to chip or final device state**

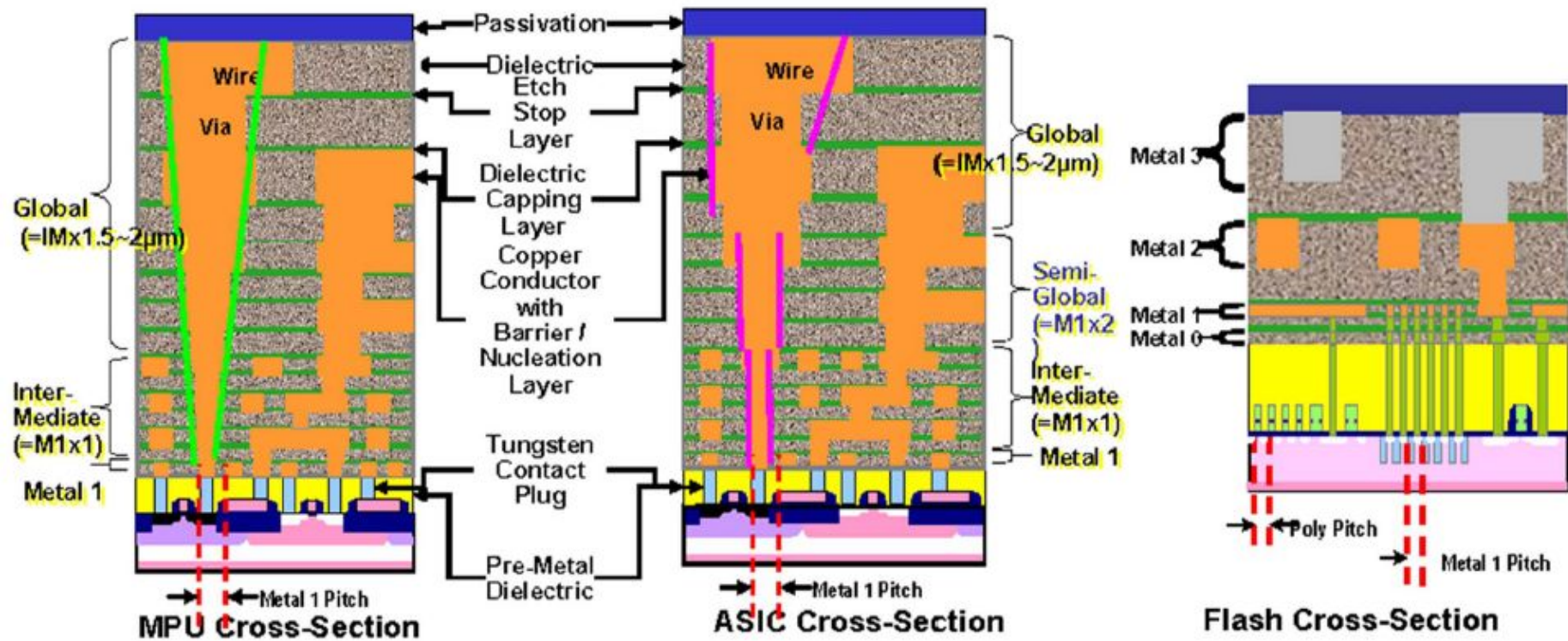
Wafer Process Flow

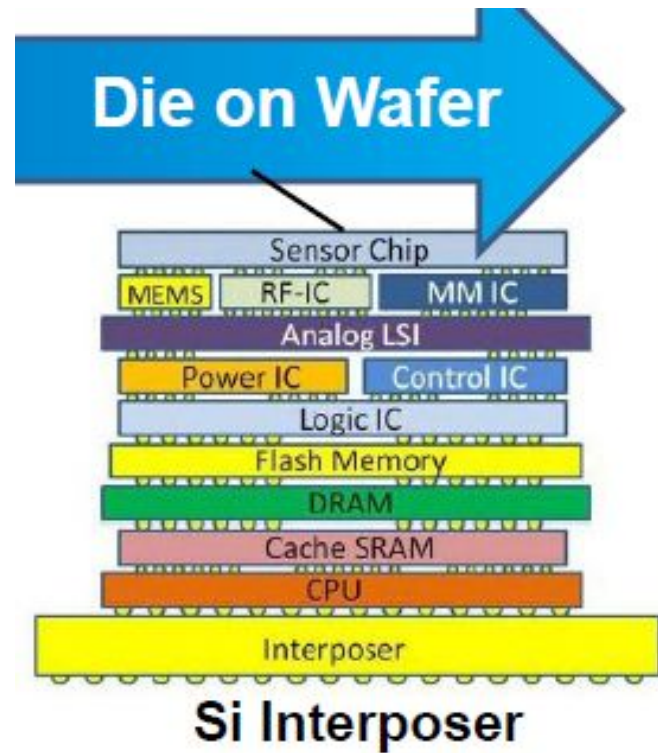
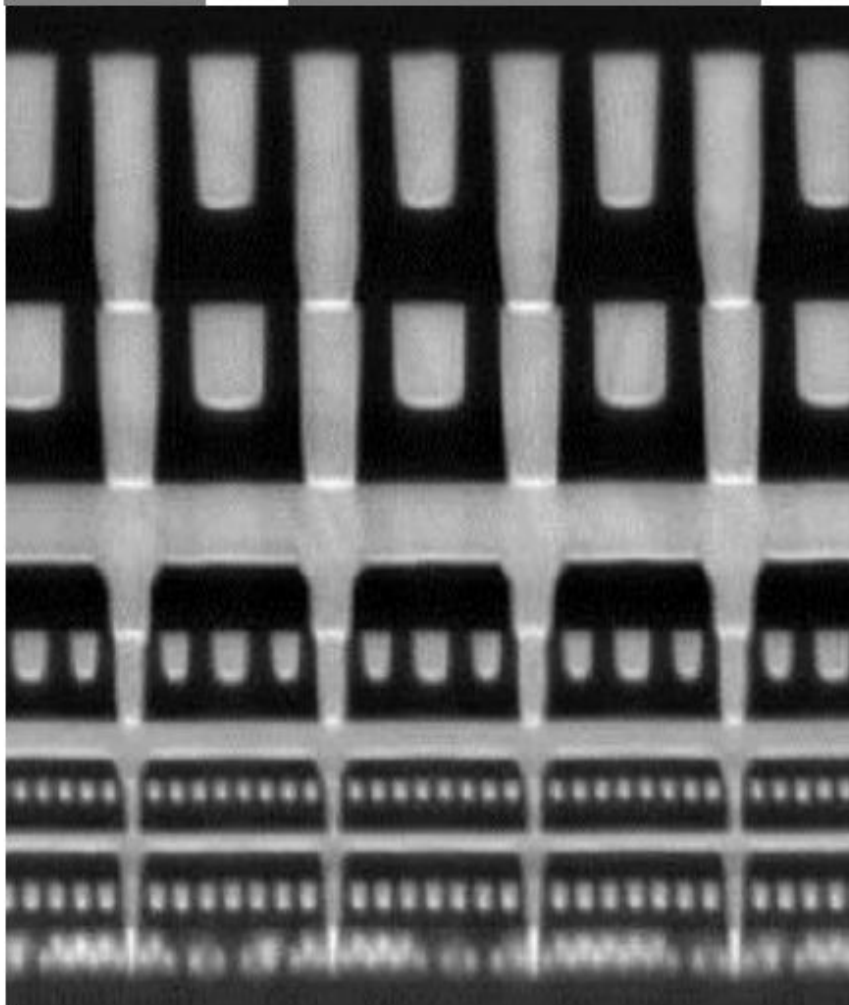






IBM 0.13um Interconnect Technology





<https://www.youtube.com/watch?v=Fxv3JoS1uY8> Zooming into a microchip quick video

And with just a microscope (optical): <https://www.youtube.com/watch?v=XEEE1uvkiH0>

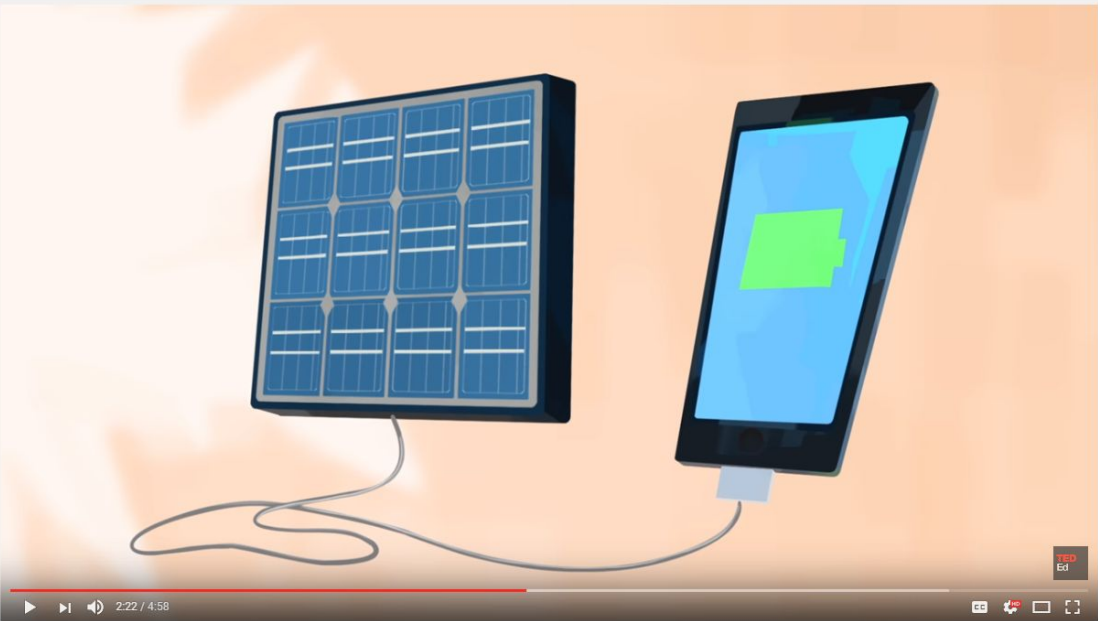
More Exciting Version



10min: <https://www.youtube.com/watch?v=fwNkg1fsqBY>

Secure | <https://www.youtube.com/watch?v=xKxrkht7CpY>

YouTube solar cells how work



2:22 / 4:58

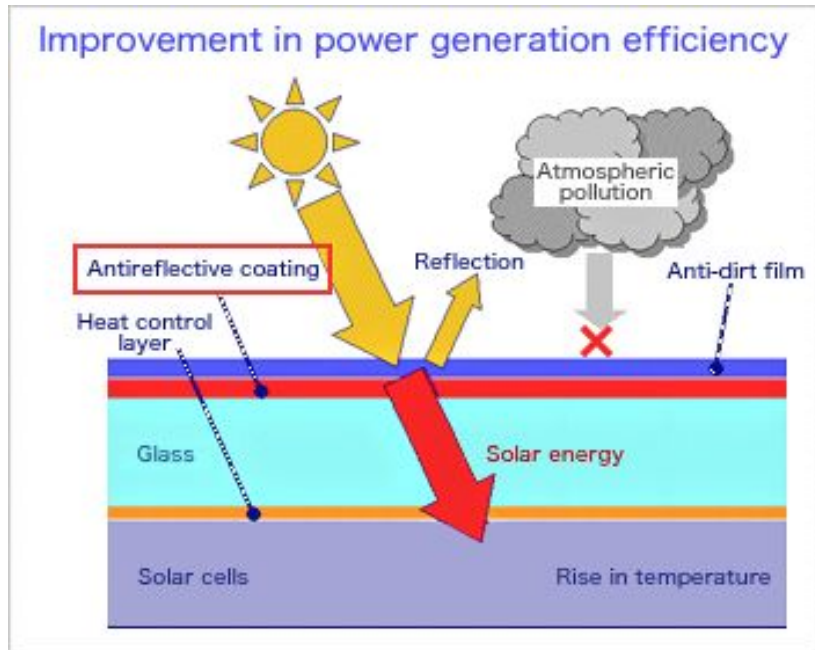
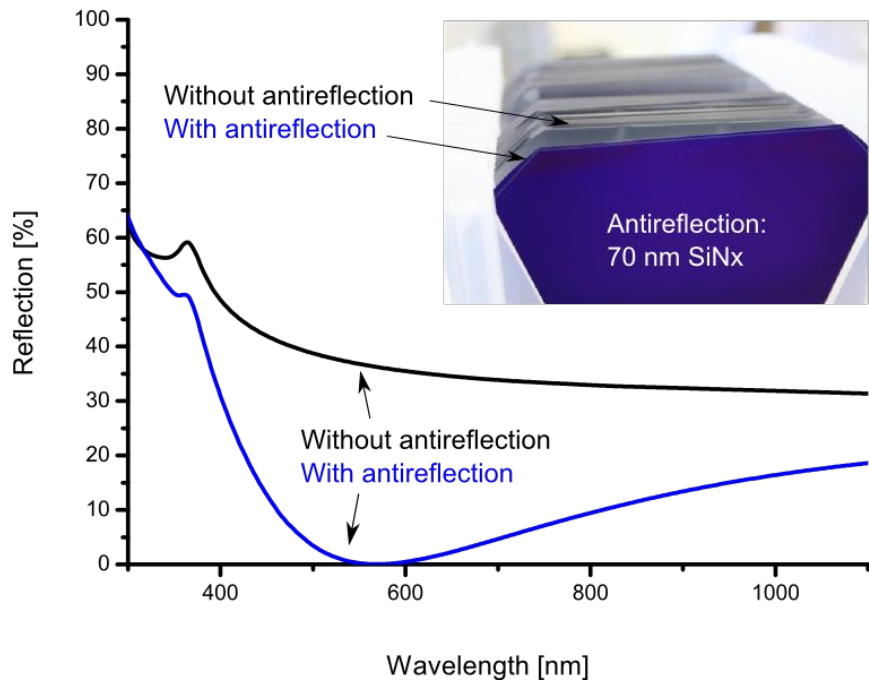
How do solar panels work? - Richard Komp

TED-Ed [Subscribe](#) 4.8M

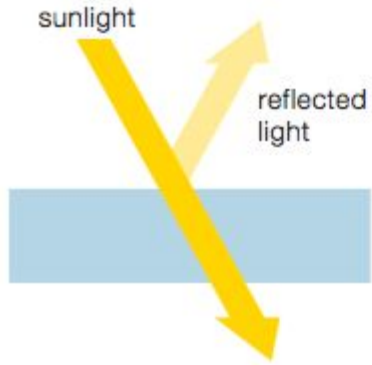
1,258,042 views

+ Add to Share More 20,367 254

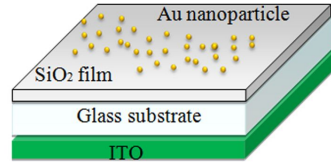
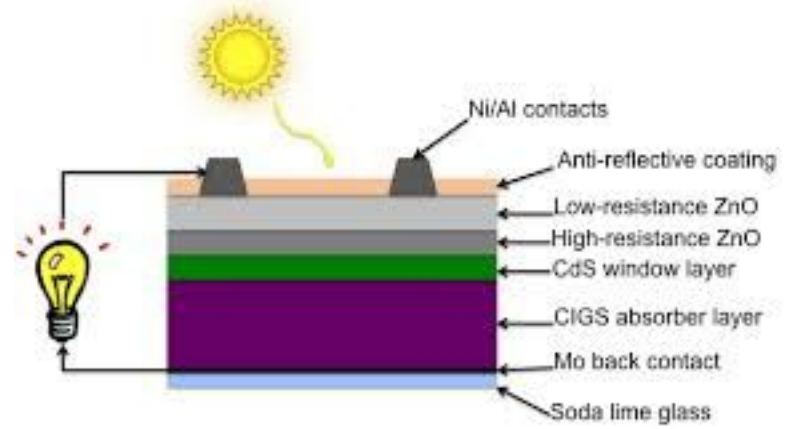
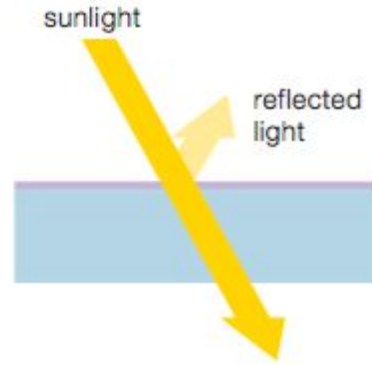
5min: <https://www.youtube.com/watch?v=xKxrkht7CpY>



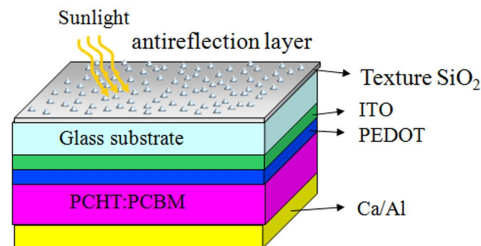
Standard Glass

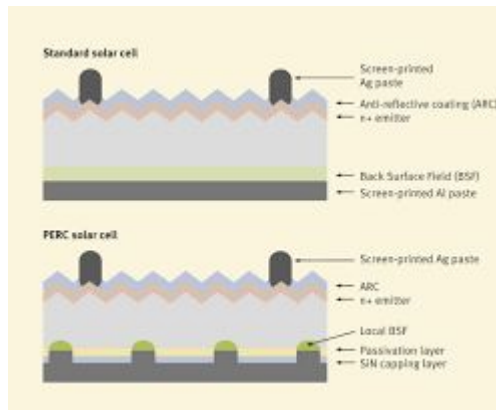
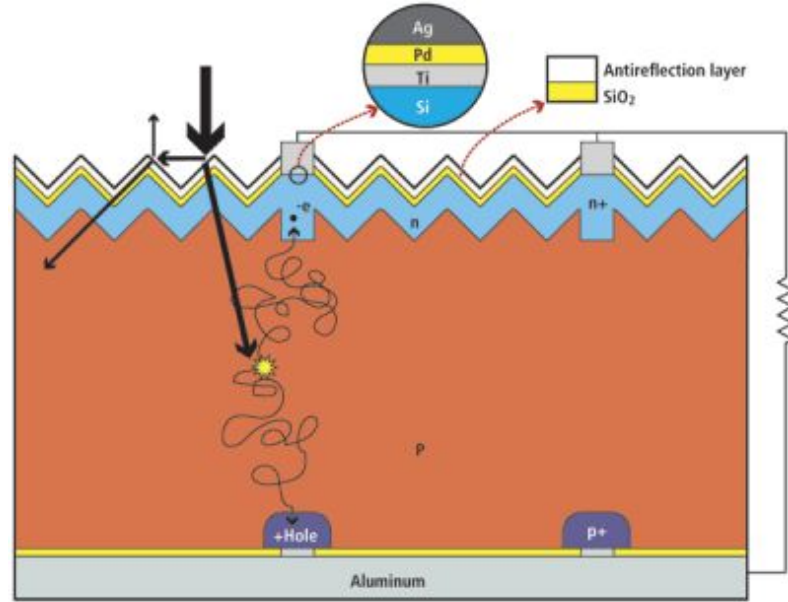
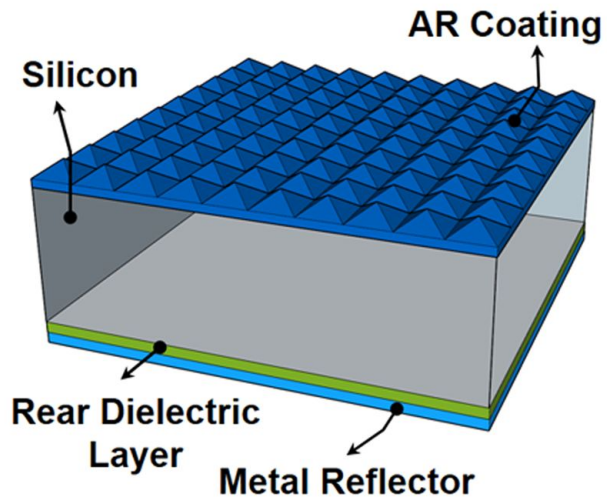


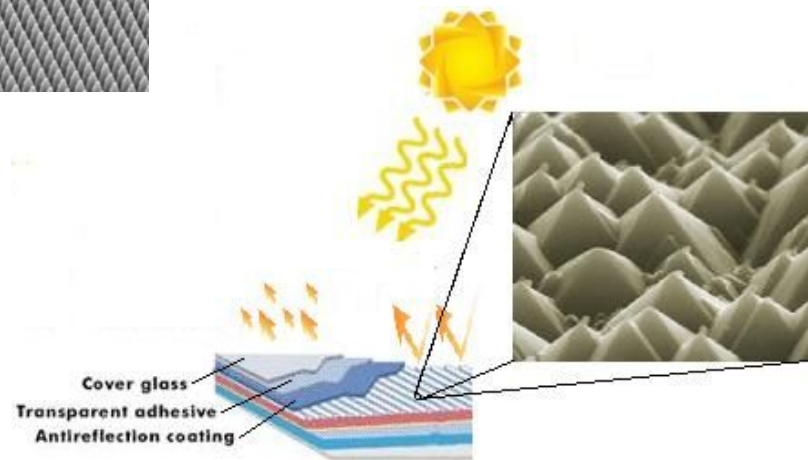
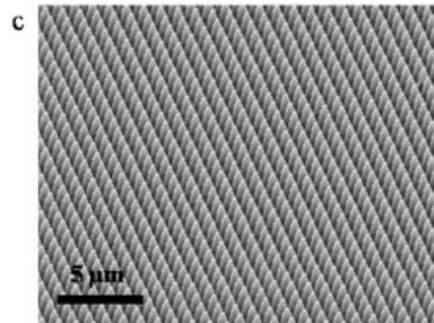
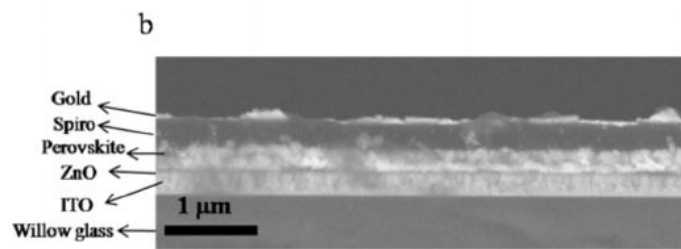
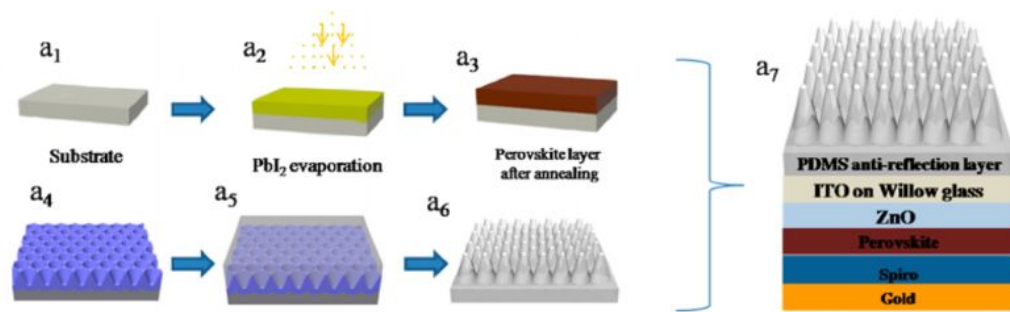
AR-treated Glass

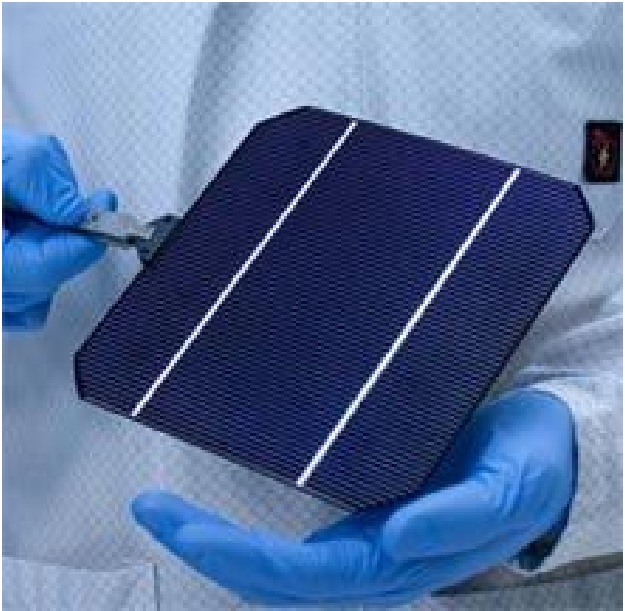


With RIE and OPV process









Intro to Deposition → **E-beam & Thermal Evaporation, Sputtering, CVD, ALD, MVD, MBE, Plating**

Intro to Photolithography → **Spin Coating, UV Exposure, Developing, E-beam Lithography, Nanoimprinting**

Intro to Etching (Wet and Dry) → **RIE, ICP-RIE, DRIE, Ion Milling, CMP, O2 Ashing**

Intro to Characterization → **AFM, SEM/FIB, TEM, Surface Profilometry, Ellipsometry, Optical Microscopes**

Intro to Back End Processes → **Wire Bonding, Dicing, Packaging**